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$2 \log(2 \cdot 10^{26}) + \log(1 \cdot 10^{15}); = 1 \cdot 2 \log 2$
 $10^{26} \cdot 1 \cdot 10^{15}; = 1 \cdot 2 \log 2 \cdot 10^{11} = \log p \cdot 2$
 10^{11} : Solve, and $d_{av} = 450$ km. E1-23
The number of atoms is given by $(1$
 $\text{kg}) / (1.00783 \cdot 1.661 \cdot 10^{-27} \text{ kg})$, or 5.974
 10^{26} atoms. E1-24 (a) $(2 \cdot 1.0 +$
 $16)u(1.661 \cdot 10^{-27} \text{ kg}) = 3.0 \cdot 10^{-26} \text{ kg}$. (b)
 $(1.4 \cdot 461021 \text{ kg}) = (3.0 \cdot 10^{-26} \text{ kg}) = 4.7 \cdot 10$

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molecules. E1-25 The coffee in Paris costs \$18.00 per kilogram, or

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m. (f) The position at $t = 2$ s is subtracted from the position at $t = 4$ s to give the

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 $= 1 \cdot 2 \log 2 \cdot 10^{11} = \log p \cdot 2 \cdot 10^{11} : \text{Solve,}$

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and $d_{av} = 450 \text{ km}$. E1-23 The number of atoms is given by $(1 \text{ kg}) / (1.00783 + 1.661 \cdot 10^{-27} \text{ kg})$, or 5:974 Halliday Physique 2 Solutions Page 4/24

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ssionals.com E25-6 Originally $F_0 = CQ^2$
 $= 0.088\text{N}$, where C is a constant. When
sphere 3 touches 1 the charge on both

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becomes $Q_0 = 2$. When sphere 3 touches sphere 2 the charge on each becomes $(Q_0 + Q_0 = 2) = 2 = 3Q_0 = 4$. The force between sphere 1 and 2 is then Page 9/26

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+100 perch² = 580 perch². We multiply this by the perch² → rood conversion factor (1 rood/40 perch²) to obtain the answer: $A_{\text{total}} = 14.5$ roods. (b) We convert our intermediate result in part (a): $A_{\text{total}} = (580 \text{ perch}^2) 16.5 \text{ ft}^2 / \text{perch}^2 = 1.58 \times 10^5 \text{ ft}^2$. Now, we use the feet → meters conversion given in Appendix D to obtain $A_{\text{total}} = 2$

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1.58×10⁵ ft² 1 m 3.281 ft

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