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response that you
require to acquire
those all needs in the
same way as having
significantly cash? Why
don't you attempt to
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a depiction of valence
electrons around the
symbol of an element.
ionic compound.
compound of cations
and anions. valence
electrons.

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Ionic Bonding. Anions
and cations are held

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together by . opposite.
charges (+ and -) Ionic
compounds are called .
salts. Formula unit -
Simplest . ratio. of
elements in an ionic
compound. Practice
with Formula Unit. ...
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Chapter 7 -
Ionic/Metallic Bonding
Vocabulary 1. Anion-

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an ion with a negative electrical charge 2.

Cation-ion with a positive electrical charge.3. Chemical Bond- is a lasting attraction between atoms, ions or molecules that enables the formation of chemical compounds.
4.

Jude Pena - Chapter 7 - Ionic_Metallic Bonding Vocabulary

...

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Bonding Answer Key

Section 7.3 - Bonding in Metals The valence electrons of metal atoms can be modeled as a sea of electrons. Metallic bonds consist of the attraction of the free-floating valence electrons for the positively charged metal ions. Metals are good conductors and malleable because of their mobile electrons.

Chapter 7 - Ionic and Metallic Bonding

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Chapter 7: Ionic and
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electrons in the highest
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compounds, atoms
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electron confi....

ionic and metallic
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bonding chapter 7

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Chapter 7 "Ionic and
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to add text 2 Section
7.1 - Ions

OBJECTIVES:-Determin
e the number of
valence electrons in an
atom of a
representative
element. 3 Section 7.1
- Ions

OBJECTIVES:-Explain
how the octet rule
applies to atoms of

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metallic and
nonmetallic elements.

4 Section 7.1 - Ions

OBJECTIVES:-Describe
how cations form. 5

Section 7.1 - Ions Chapter 7 “Ionic and Metallic Bonding”

Section 7.3 Bonding in
Metals OBJECTIVES:

-Explain the
importance of alloys.

45. Metallic Bonds
are... How metal atoms
are held together in
the solid. Metals hold

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on to their valence electrons very weakly. Think of them as positive ions (cations) floating in a sea of electrons: Fig. 7.12, p.201. 46.

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together by ionic
bonds. composed of
anions and cations. all
of the above. Tags:
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30 seconds . Q. Which
of these is NOT a
characteristic of most
ionic ...

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occupied energy level
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compounds, atoms
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Section 7.1 - Ions

OBJECTIVES: -

Determine the number
of valence electrons in
an atom of a
representative

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BONDING AND
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Ions.
Essential Understanding
Ions form when atoms

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gain or lose valence electrons, becoming electrically charged.

Lesson Summary.

Valence

Electrons

Valence electrons are the electrons in the outermost occupied energy level and are involved in ion formation. For a representative element, the group number equals the ...

BONDING AND

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Chapter 7 - Chemical Bonding
7.1 Ionic Bonding
Octet rule: In forming compounds atoms lose, gain or share electrons to attain a noble gas configuration with 8 electrons in their outer shell (s^2p^6), except H and He want 2 outer electrons ($1s^2$).
Basically atoms want to be like the noble gases which are stable and happy atoms.

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Compounds and Metals

Section 7.1.1 Figure
7.1 As carbon dioxide

dis-solves in ocean
water, carbonate ions

are produced. Coral
polyps capture these

car-bonate ions,
producing crystals of

calcium carbonate,
which they secrete as
an exo-skeleton. Over

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time, the coral reef forms. A coral reef is a complex habitat that

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